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Important One Liner for All Agriculture Competitive Examination

- NH_3 is a cheapest source of **N fertilizer**.
- 1 kg NH_3 needs 1.8 kg lime to **neutralize acidity** developed.
- **Ammonium sulphate** is manufactured by the
 1. Coke oven gas process – Bituminous coal is subjected to destructive distillation. It contains 10% NH_3 by volume.
 2. 85% of $(\text{NH}_4)_2 \text{SO}_4$ manufactured in India is by “Gypsum process”.
- **$(\text{NH}_4)_2 \text{SO}_4$** is white crystalline salt, but commercially vary from yellow to gray due to impurities like ferric cyanide, arsenous sulphides (or) traces of tar
- “**Thiocyanates**” present are toxic to plants. So the $(\text{NH}_4)_2 \text{SO}_4$ are periodically checked.
- 110 kg of CaCO_3 is needed to neutralize the acidity created by 100 kg of **$(\text{NH}_4)_2 \text{SO}_4$** .
- 1kg of N applied as ammonium sulphate requires 5.1kg of lime for **neutralization**.
- **NH_4Cl** is very popular in Japan and S.E. Asia. It is prepared by
 1. Direct neutralization process.
 2. Dual salt (or) modified Solvay process. Here NH_4Cl is salted out by NaCl using the principle of “common ion effect”.
- **CAN** is a mixture of NH_4NO_3 and CaCO_3 .
- Fine granules of CAN are coated with soap stone powder i.e. **MgSiO_3** which improves the free flowing character of the fertilizer and it acts as a conditioner.
- “Urea (or) carbamide” has N in organic form, **non-ionic** form (or) amide form.
- 82% of N requirement of the crops in India is by **urea**.

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