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- When temperature exceeds 100°C, urea produces a toxic product called **“biuret”**. Biuret has less amount of ‘N’. $\text{NH}_2 - \text{CO} - \text{NH} - \text{CO} - \text{NH}_2$.
- The amount of **biuret content** in urea should not exceed **1.5%**; when urea is being used as a feed to cattle the biuret content should be **less than 0.5%**.
- When soil is acidic, **ammonical fertilizers** should not be recommended as they increase **acidity**.
- For **paddy ammonical fertilizers** are preferable as paddy takes N in the form of **NH_4^+**
- **PO_4^{3-}** which is unavailable form after being added to soils. **H_2PO_4^- & HPO_4^{2-}** are available forms.
- Rock phosphate formed from primary minerals is **“magmatic origin”** and that formed due to teeth and bones is called **“organogenic origin”**.
- **“Morocco”** occupies first place in deposits of P. The rock phosphate here is of organogenic origin.
- The rock Phosphate of N. Africa is very soft and finely crystalline and is called **“Galsa phosphate”**.
- In USA rock phosphate is present in the form of **“pebbles”**.
- In USSR rock phosphate is very hard and called **“Kolaphosphats”**.
- **“Rock phosphate”** is the basic raw material for manufacturing any phosphatic fertilizer.
- Rock phosphate when treated with H_2SO_4 give SSP; when treated with H_3PO_4 gives **‘TSP’**.
- H_3PO_4 can directly used as a **liquid fertilizer**.
- **“Soft earth”** of R.P is best for acidic soils, for grasslands, pastures and orchards where the ‘P’ requirement is needed over a long period of time.

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